```
PART I- TRANSPORT PROCESSES: MOMENTUM, HEAT, AND MASS
0.2 - 11
            X = 353.2 = °C+ 273.15
Eq. (1.3-4)
                                      °C= 80.05= 80°C
                                                176°F
Eq. (1.3-2)
           °C=80.0=(1/1.8)(°F-32)
                                                636°R
                                           * R =
           OR = "F+ 460= 176+460
Eg.(1.3-3)
(1, 2-2)
 Eq. (1.3-2) °C = (1/1.8)(°F-32)=(1/1.8)(155-32) = 68.33°C
 Eq. (1.3-4) K = °C + 273,15 = 68.33+273,15 =
                                             341.5 K
 Eq. (1.3-3) °R = °F + 460 = 155 + 460 =
                                             615 °R
0.3-0
         MW(02)= 32.00, MW(N2)= 28.02
  MW (air) = 0.21(32.00)+0.79(28.02)=28.9 deg/deg mod
0.3-22
    200 + 02 -> 2002
                             moles CO = 56.0 kg
  MW 28.01 32,00
                    44.01
 moles 02 = 1.00, 1.00(32.00) = 32.0 kg 02 2.00(44.0)= 88.0 kg CO2
(1.3 - 3)
    Eas
                            a mol mol free
                   MW
                                         0.1648 molfrer
      N2 20 28.02
                            0.7/38
      02
             83
                   3 2.00
                            3.5728
                                         0.5 990
     CO2 45
                   44.01
                           1.0225
                                         0.2362
                                         1.0000
                            4.33015 mol
     Jotal 1489
   average mol wt = 1489 + 33019 mol =
```

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