

Reminders:

1. In a neutral atom the number of protons equals the number of electrons.
2. An atom can NEVER gain or lose protons
3. The number of protons equals the atomic number

Important!

NAME: _____

Answer Key

Ion Practice Set

1. What is an ion?

a charged ion - it can be positively charged (cation) or negatively charged (anion).

2. What does the number next to the ions signify?
the charge of the ion

Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED
ex	Fluorine	F ⁻	9	10	gained one
1	Iodine	I ⁻	53	54	gained one
2	Sulfur	S ²⁻	16	16+2 = 18	gained two
3	potassium	K ⁺	19	19-1 = 18	lost one
4	Calcium	Ca ²⁺	20	20-2 = 18	lost two
5	Bromine	Br ⁻	35	36	gained one
6	Strontium	Sr ²⁺	38	38-2 = 36	lost two
7	Hydrogen	H ⁺	1	1-1 = 0	lost one
8	Oxygen	O ²⁻	8	8+2 = 10	gained two
9	Magnesium	Mg ²⁺	12	12-2 = 10	lost two
10	aluminum	Al ³⁺	13	10	lost three
11	Selenium	Se ²⁻	34	36	gained two
12	Hydrogen	H ⁺	1	1+1 = 2	gained one
13	lithium	Li ⁺	3	3-1 = 2	lost one
14	Rubidium	Rb ⁺	37	37-1 = 36	lost one
15	Chlorine	Cl ⁻	17	18	gained one

this means it has an extra electron

this means it has 2 more protons than electrons, so 2 electrons were lost!

Pumping Iron Practice Set Answer Key

Chandresh Agrawal, Nandini Books

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Table of Contents Pumping Iron Practice Set Answer Key

1. Understanding the eBook Pumping Iron Practice Set Answer Key
 - The Rise of Digital Reading Pumping Iron Practice Set Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pumping Iron Practice Set Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pumping Iron Practice Set Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pumping Iron Practice Set Answer Key
 - Personalized Recommendations
 - Pumping Iron Practice Set Answer Key User Reviews and Ratings
 - Pumping Iron Practice Set Answer Key and Bestseller Lists
5. Accessing Pumping Iron Practice Set Answer Key Free and Paid eBooks
 - Pumping Iron Practice Set Answer Key Public Domain eBooks
 - Pumping Iron Practice Set Answer Key eBook Subscription Services
 - Pumping Iron Practice Set Answer Key Budget-Friendly Options

6. Navigating Pumping Iron Practice Set Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Pumping Iron Practice Set Answer Key Compatibility with Devices
 - Pumping Iron Practice Set Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pumping Iron Practice Set Answer Key
 - Highlighting and Note-Taking Pumping Iron Practice Set Answer Key
 - Interactive Elements Pumping Iron Practice Set Answer Key
8. Staying Engaged with Pumping Iron Practice Set Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pumping Iron Practice Set Answer Key
9. Balancing eBooks and Physical Books Pumping Iron Practice Set Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pumping Iron Practice Set Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pumping Iron Practice Set Answer Key
 - Setting Reading Goals Pumping Iron Practice Set Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pumping Iron Practice Set Answer Key
 - Fact-Checking eBook Content of Pumping Iron Practice Set Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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