

Preparation of Solution

Molar Solutions

What do you know?

Mass of Solute (g)

Final Volume of Solution (L)

$$\text{Mass of Solute (g)} = \frac{1}{\text{Molar Mass } \left(\frac{\text{g}}{\text{mole}}\right)} \times \frac{1}{\text{Final Volume of solution (L)}} = \text{Molarity (M or } \frac{\text{mole}}{\text{L}})$$

What do you want to know?

Mass of Solute (g)

Final Volume of Solution (L)

$$\text{Molarity (M or } \frac{\text{mole}}{\text{L}}) \times \text{Final Volume of solution (L)} \times \text{Molar Mass } \left(\frac{\text{g}}{\text{mole}}\right) = \text{Mass of Solute (g)}$$

What do you want to know?

Mass of Solute (g)

Final Volume of Solution (L)

$$\text{Mass of Solute (g)} = \frac{1}{\text{Molar Mass } \left(\frac{\text{g}}{\text{mole}}\right)} \times \frac{1}{\text{Molarity (M or } \frac{\text{mole}}{\text{L}})} = \text{Final Volume of Solution (L)}$$

What do you know?

Mass of Solute (g)

Final Volume of Solution (L)

What do you want to know?

Final Volume of Solution (L)

Percent (%) Solutions

Mass/Mass Percent Concentration Solution - All percentages should be in decimal form

What do you know?

Mass of Solute (g)

Mass of Solvent (g)

$$\frac{\text{Mass of Solute (g)}}{\text{Mass of Solute (g)} + \text{Mass of Solvent (g)}} = \text{Percent Concentration of Solution (\%)}$$

What do you want to know?

Percent Concentration of Solution (%)

What do you know?

Mass of Solute (g)

Percent Concentration of Solution (%)

$$\frac{\text{Mass of Solute (g)}}{\text{Percent Concentration of Solution (\%)}} = \text{Mass of Solvent (g)} + \text{Mass of Solute (g)}$$

What do you want to know?

Mass of Solvent (g)

What do you know?

Mass of Solvent (g)

Percent Concentration of Solution (%)

$$\frac{\text{Percent Concentration of Solution (\%)}}{1 - \text{Percent Concentration of Solution (\%)}} = \text{Mass of Solute (g)}$$

What do you want to know?

Mass of Solute (g)

Mass/Volume Percent Solutions - All percentages should be in decimal form

What do you know?

Mass of Solute (g)

Final Volume of Solution (mL)

$$\frac{\text{Mass of Solute (g)}}{\text{Final Volume of Solution (mL)}} = \text{Percent Concentration of Solution (\%)}$$

What do you want to know?

Percent Concentration of Solution (%)

What do you know?

Mass of Solute (g)

Percent Concentration of Solution (%)

$$\frac{\text{Mass of Solute (g)}}{\text{Percent Concentration of Solution (\%)}} = \text{Final Volume of Solution (mL)}$$

What do you want to know?

Final Volume of Solution (mL)

What do you know?

Final Volume of Solution (mL)

Percent Concentration of Solution (%)

$$\text{Final Volume of Solution (mL)} \times \text{Percent Concentration of Solution (\%)} = \text{Mass of Solute (g)}$$

What do you want to know?

Mass of Solute (g)

Dilutions

Molar Solution:

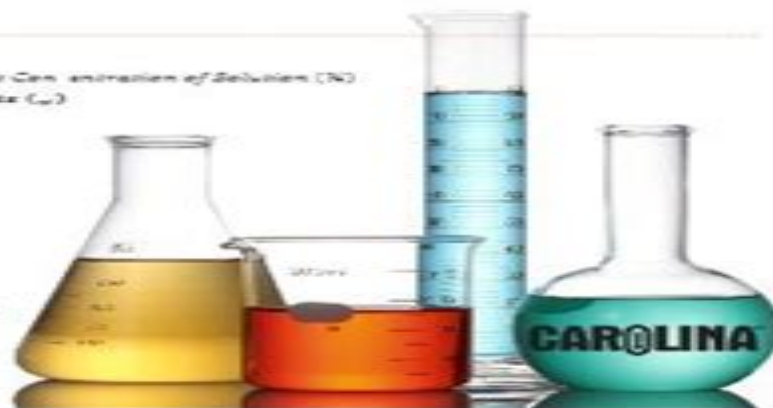
Molarity of Initial Solution $\left(\frac{\text{mole}}{\text{L}}\right) \times \text{Volume of Initial Solution (L)} =$

Molarity of Final Solution $\left(\frac{\text{mole}}{\text{L}}\right) \times \text{Volume of Final Solution (L)}$

Percent Concentration Solutions

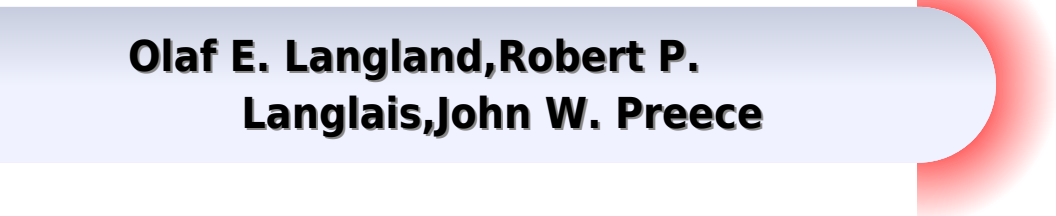
Percent Concentration of Initial Solution $\left(\frac{\text{g}}{\text{mL}}\right) \times \text{Volume of Initial Solution (L)} =$

Percent Concentration of Final Solution $\left(\frac{\text{g}}{\text{mL}}\right) \times \text{Volume of Final Solution (L)}$



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**Olaf E. Langland, Robert P.
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manufakturen so zum beispiel glas für iittala und porzellanserien für rosenthal von ihm stammt eine

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web noch bis anfang oktober 2016 ist im grassi museum für angewandte kunst in leipzig die ausstellung tapio Wirkkala finnisches design glas und silber zu sehen das glasmuseum riihimäki hat anlässlich des 100 geburtstages von tapio Wirkkala die ausstellung zusammengestellt

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web jun 2 2015 am 3 juni 2015 startete im design forum finnland in helsinki die verkaufsausstellung tapio Wirkkala the colours of venice bei der objekte des finnischen designers für die italienische glasmanufaktur venini gehandelt wurden zur ausstellung Wirkkala revisited mit unikaten massenwaren und 3 d drucken lud das

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