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 $\text{baso}_4 + \text{sba}_2 \rightleftharpoons \text{aq} + \text{so}_4^{2-} + \text{aq} + \text{ag}_2\text{so}_4 + \text{s} + 2\text{ag} + \text{aq} + \text{so}_4^{2-} + 4\text{aq} + \text{xal} + \text{oh}^- + 3\text{s} + \text{al} + 3\text{aq} + 3\text{oh}^- + \text{aq} + \text{xpb} + \text{oh}^- + \text{cl} + \text{s} + \text{pb} + 2\text{aq} + \text{oh}^- + \text{aq} + \text{cl} + \text{aq} + \text{x}$

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$\text{Ca}^{2+}(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{CaCO}_3(\text{s})$

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