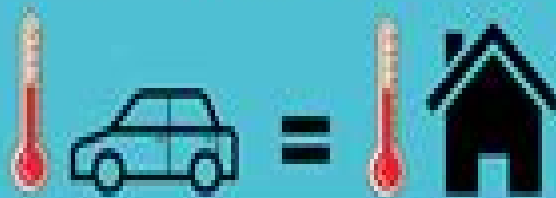


Laws of Thermodynamics

Zeroth Law

Defines temperature equality: if two bodies are each with a third, they're in equilibrium.



First Law

Energy remains constant; it transforms but cannot be created or destroyed.



Total Energy
Equal



Second Law

Entropy increases in isolated systems; heat flows from hot to cold.



Third Law

Entropy approaches a constant value as temperature approaches absolute zero.



Principles Thermodynamics

Robert Röntgen



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