

Exercise 1: Qualitative Anion Tests

Data Table 1. Anion Confirmation Tests.

Chemical	Anion	Addition of AgNO_3	Addition of HCl
<u>NaBr</u>	Bromide	Milky like precipitate formed	No initial changes, solids began to fall out after an hour
Na_2CO_3	Carbonate	Precipitate formed, white in color, and began to separate	Mixture cleared, precipitate clumped together, small bubble could be seen on the bottom
NaCl	Chloride	Mixture turned completely white, small amount of clumping from precipitate being formed	Solids pulled out of the mixture; no other reaction occurred
<u>NaI</u>	Iodide	Mixture turned opalescent; precipitate formed	Colors in mixture faded slightly, no other reaction
Na_2SO_4	Sulfate	No observable reaction	Mixture turned white
Na_2S	Sulfide	Mixture turned brown black in color, solids forms and sank to bottom.	No noticeable changes, did notice smell of rotten eggs,
Unknown #103	Carbonate	Solids pulled out of mixture	Bubble formed
Unknown #104	Sulfide	Turned brown black, solids separated and sank	Rotten eggs smell

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Emil J. Margolis



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