

The Periodic Table of Elements

In 1869 Dmitri Mendeleev organized the elements in a table based on the properties of the elements.

Groups
(columns)



The elements in each group have the same number of valence electrons

1st Period (2 shells)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
2nd Period (2 shells)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
3rd Period (3 shells)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
4th Period (4 shells)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18

Periods
(rows)

The elements in each period have the same number of shells

Today, 114 elements are arranged by atomic number in the periodic table. The atomic number is based on the number of protons in an element.

How to Read the Periodic Table

Atomic Number
of protons in the nucleus
Same # of electrons in the nucleus

6

C

Carbon

12.0

1 or 2 letters that represent the element

Element name

Atomic Mass
the total mass of the atom of that element

Mass number
sum of the numbers of protons and neutrons in a specific isotope
(12)

Protons ●●●●●●

Neutrons ●●●●●●

Elements are listed in order of increasing atomic number.

Periodic Table Of Elements Lab Activities

Pam Walker, Elaine Wood



Periodic Table Of Elements Lab Activities:

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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