

Study Guide – Unit 1 Periodic Table **ANSWERS**

Structure of an atom

1. List the three subatomic particles which make up an atom. Include their location, charge, and atomic mass.

a. Electron	Negative charge	Electron Cloud
b. Proton	Positive charge	Nucleus
c. Neutron	Neutral charge	Nucleus
2. In a neutral atom, which two subatomic particles are equal in number to each other?
Protons and Electrons
3. What does the atomic number of an element indicate?
The number of protons in the atom of that element
4. What is an isotope?
Atoms with the same number of protons, but different masses due to a different number of neutrons
5. Will two different atoms of carbon always have the same number of protons?
Yes

Elemente mit Kern

13
Al
26.982
Aluminum

1. What are the number of protons, neutrons, and electrons in an atom of aluminum?
- Protons = 13
 - Neutrons = 14
 - Electrons = 13

Bohr Model

- How do you know how many energy levels (orbits) a Bohr model will have?
a. By the period the element is located in
- How many electrons can each energy level hold?
1st - 2 2nd - 8 3rd - 18

The Periodic Table

1. How is the periodic table arranged? **Increasing atomic number**
2. What are the vertical columns $\updownarrow$ called? **Groups or Families**
3. What are the horizontal rows $\leftrightarrow$ called? **Periods**
4. Why are elements put into groups? **Same number of valence electrons**
5. What is a valence electron? **Number of electrons on the outer energy level (shell)**
6. Know how many valence electrons each group has
7. Which group of metals are highly reactive and why? **Alkali Metals. Only 1 ve to lose**
8. Which group of non-metals are highly reactive and why? **Halogens. Only 1 ve to gain**
9. Name some physical properties of metals: **luster, ductile, malleable, conductor of heat and electricity**
10. Name some physical properties of non metals: **no luster, brittle, not a good conductor**
11. What are metalloids and where are they found on the periodic table? **Share properties of metals and metalloids. On zig-zag line starting at Boron.**

Periodic Table Study Guide Answers

Chao Zhang

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located on the right side of the light blue bar.

Periodic Table Study Guide Answers:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Know Your 'O' Level Chemistry - A Study Guide , Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 **Class 9 Chemistry Quiz PDF: Questions and Answers Download | 9th Grade Chemistry Quizzes Book** Arshad Iqbal, The Book Class 9 Chemistry Quiz Questions and Answers PDF Download 9th Grade Chemistry Quiz PDF Book Chemistry Interview Questions for Teachers Freshers Chapter 1 8 Practice Tests Class 9 Chemistry Textbook Questions to Ask in Job Interview includes revision guide for problem solving with hundreds of solved questions Class 9 Chemistry Interview Questions and Answers PDF covers basic concepts analytical and practical assessment tests Class 9 Chemistry Quiz Questions PDF Book helps to practice test questions from exam prep notes The e Book Class 9 Chemistry job assessment tests with answers includes revision guide with verbal quantitative and analytical past papers solved tests Class 9 Chemistry Quiz Questions and Answers PDF Download a book covers solved common questions and answers on chapters Chemical reactivity electrochemistry fundamentals of chemistry periodic table and periodicity physical states of matter solutions structure of atoms structure of molecules tests for school and college revision guide Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The Book Class 9 Chemistry Interview Questions Chapter 1 8 PDF includes high school question papers to review practice tests for exams Class 9 Chemistry Practice Tests a textbook s revision guide with chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam 9th Grade Chemistry Questions Bank Chapter 1 8 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Chemical Reactivity Questions Chapter 2 Electrochemistry Questions Chapter 3 Fundamentals of Chemistry Questions Chapter 4 Periodic Table and Periodicity Questions Chapter 5 Physical States of Matter Questions Chapter 6 Solutions Questions Chapter 7 Structure of Atoms Questions Chapter 8 Structure of Molecules Questions The e Book Chemical Reactivity quiz questions PDF chapter 1 test to download interview questions Metals and non metals The e Book Electrochemistry quiz questions PDF chapter 2 test to download interview questions Corrosion and prevention electrochemical cells electrochemical industries oxidation and reduction oxidation reduction and reactions oxidation states oxidizing and reducing agents The e Book Fundamentals of

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