

The Periodic Table of Elements

In 1869 Dmitri Mendeleev organized the elements in a table based on the properties of the elements.

Groups



The elements in each **group** have the same number of **valence electrons**.

1st Period (2 elements)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
2nd Period (8 elements)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
3rd Period (8 elements)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
4th Period (18 elements)	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18

Periods

The elements in each **period** have the same number of **shells**.

Today, 114 elements are arranged by atomic number in the periodic table. The atomic number is based on the number of protons in an element.

How to Read the Periodic Table

Atomic Number
of protons in the nucleus
Same # of electrons in the nucleus

6

C

Carbon

12.0

1 or 2 letters that represent the element

Element name

Atomic Mass
the total mass of the atom of that element

Mass number
sum of the numbers of protons and neutrons in a specific isotope
(12)

Protons ●●●●●●

Neutrons ●●●●●●

Elements are listed in order of increasing atomic number.

Periodic Table Of Elements Lab Activities

**K. J. Schulz, John H. DeYoung, Robert R.
Seal, Dwight C. Bradley**



Periodic Table Of Elements Lab Activities:

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Labster Virtual Lab Experiments: Basic Biochemistry Aaron Gardner, Wilko Duprez, Sarah Stauffer, Dewi Ayu Kencana Ungu, Frederik Clauson-Kaas, 2019-04-01 This textbook helps you to prepare for your next exams and practical courses by combining theory with virtual lab simulations The Labster Virtual Lab Experiments series gives you a unique opportunity to apply your newly acquired knowledge in a learning game that simulates exciting laboratory experiments Try out different techniques and work with machines that you otherwise wouldn't have access to In this book you'll learn the fundamental concepts of basic biochemistry focusing on Ionic and Covalent Bonds Introduction to Biological Macromolecules Carbohydrates Enzyme Kinetics In each chapter you'll be introduced to one virtual lab simulation and a true to life challenge Following a theory section you'll be able to play the relevant simulation that includes quiz questions to reinforce your understanding of the covered topics 3D animations will show you molecular processes not otherwise visible to the human eye If you have purchased a printed copy of this book you get free access to five simulations for the duration of six months If you're using the e book version you can sign up and buy access to the simulations at www.labster.com springer If you like this book try out other topics in this series including Basic Biology Basic Genetics and Genetics of Human Diseases Please note that the simulations in the book are not virtual reality VR but 2D virtual experiments

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Nuclear Science Abstracts, 1976

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The Art of Teaching Science Jack Hassard, Michael Dias, 2013-07-04 The Art of Teaching Science emphasizes a humanistic experiential and constructivist approach to teaching and learning and integrates a wide variety of pedagogical learning tools These tools involve inquiry and experimentation reflection through writing and discussion as well

as experiences with students science curriculum and pedagogy Becoming a science teacher is a creative process and this innovative textbook encourages students to construct ideas about science teaching through their interactions with peers professionals and instructors and through hands on minds on activities designed to foster a collaborative thoughtful learning environment

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