

Name:

Date:

Student Exploration: Moles

Directions: Follow the instructions to explore the Gizmo and respond to the questions and prompts in the boxes.

Vocabulary: atomic mass, Avogadro constant, conversion factor, dimensional analysis, mole, molar mass, molecular mass, scientific notation, significant figures, unified atomic mass unit

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- In the image to the right, note a dozen eggs, a dozen donuts and a dozen roses. How many of each item do you have?

- Would a dozen of each object have the same mass? _____

- Suppose you have a dozen carbon atoms, a dozen gold atoms, and a dozen iron atoms. Even though you have the same number of each, would you expect them all to have the same mass? Explain.



Gizmo Warm-up

When counting roses, donuts, or eggs, a dozen is 12. If you are counting atoms, how many is a dozen? (It is not 12.) How many atoms will you find about a unit used to count atoms?

On the **AVOGADRO CONSTANT** tab, place the copper (Cu) atom on the nano-balance on the left, which will show the average atomic mass of copper rather than the mass of a single copper atom.

- What is the average mass of a copper atom? _____

The unit "u" refers to **unified atomic mass units**. A single proton or neutron has a mass of approximately one atomic mass unit. (Officially, 1 u is one-twelfth the mass of a C-12 atom.)

- To gain an idea as to how many atoms are in a gram or so of copper, use the larger balance on the right. Press **Add atoms** to put a scoop of atoms in the weighing dish, and keep adding until the balance registers between 1 and 2 grams. If you don't seem to be making much progress, adjust the exponent using the slider, which will make the scoop size bigger. How many atoms did you need to add? _____



Mole Lab Student Guide Answer Key

Stephanie Kuligowski, Karen Kroeter

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author names.

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Laboratory Guide to Enzymology Geoffrey A. Holdgate, Antonia Turberville, Alice Lanne, 2024-03-05

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