

Periodic Table of the Elements

This table gives information about the chemical elements. Elements are grouped into eight classes according to their properties. Each class is shown in a different color. Hydrogen does not belong to any one class.

Atomic number — 6
Chemical symbol — C
Element name — Carbon
Number of electrons in each shell — 2, 4
Atomic weight — 12.011
(3) indicates atomic weight of the most stable isotope

1 H Hydrogen 1.00794	2 He Helium 4.002602											13 B Boron 10.811	14 C Carbon 12.011	15 N Nitrogen 14.00643	16 O Oxygen 15.9994	17 F Fluorine 18.9984032	18 Ne Neon 20.1797
3 Li Lithium 6.941	4 Be Beryllium 9.012182											19 Al Aluminum 26.9815385	20 Si Silicon 28.0855	21 P Phosphorus 30.973762	22 S Sulfur 32.06	23 Cl Chlorine 35.4527	24 Ar Argon 39.948
5 Na Sodium 22.98976928	6 Mg Magnesium 24.304	7 Sc Scandium 44.955912	8 Ti Titanium 47.867	9 V Vanadium 50.9415	10 Cr Chromium 51.9961	11 Mn Manganese 54.938045	12 Fe Iron 55.845	13 Co Cobalt 58.933195	14 Ni Nickel 58.6934	15 Cu Copper 63.546	16 Zn Zinc 65.38	17 Ga Gallium 69.723	18 Ge Germanium 72.64	19 As Arsenic 74.9216	20 Se Selenium 78.96	21 Br Bromine 79.904	22 Kr Krypton 83.80
11 K Potassium 39.0983	12 Ca Calcium 40.078	13 Sc Scandium 44.955912	14 Ti Titanium 47.867	15 V Vanadium 50.9415	16 Cr Chromium 51.9961	17 Mn Manganese 54.938045	18 Fe Iron 55.845	19 Co Cobalt 58.933195	20 Ni Nickel 58.6934	21 Cu Copper 63.546	22 Zn Zinc 65.38	23 Ga Gallium 69.723	24 Ge Germanium 72.64	25 As Arsenic 74.9216	26 Se Selenium 78.96	27 Br Bromine 79.904	28 Kr Krypton 83.80
19 K Potassium 39.0983	20 Ca Calcium 40.078	21 Sc Scandium 44.955912	22 Ti Titanium 47.867	23 V Vanadium 50.9415	24 Cr Chromium 51.9961	25 Mn Manganese 54.938045	26 Fe Iron 55.845	27 Co Cobalt 58.933195	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.64	33 As Arsenic 74.9216	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80
37 Rb Rubidium 85.4678	38 Sr Strontium 87.62	39 Y Yttrium 88.90584	40 Zr Zirconium 91.224	41 Nb Niobium 92.90638	42 Mo Molybdenum 95.94	43 Tc Technetium 98	44 Ru Ruthenium 101.07	45 Rh Rhodium 101.07	46 Pd Palladium 106.3675	47 Ag Silver 107.8682	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.90547	54 Xe Xenon 131.29
55 Cs Cesium 132.90545196	56 Ba Barium 137.327	57 La Lanthanum 138.90547	58 Ce Cerium 140.12	59 Pr Praseodymium 140.90766	60 Nd Neodymium 144.24	61 Pm Promethium 144.9127	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92534	66 Dy Dysprosium 162.5	67 Ho Holmium 164.93033	68 Er Erbium 167.257	69 Tm Thulium 168.93048	70 Yb Ytterbium 173.05448	71 Lu Lutetium 174.967	
87 Fr Francium 223	88 Ra Radium 226	89 Ac Actinium 227	90 Th Thorium 232.0377	91 Pa Protactinium 231.036889	92 U Uranium 238.02891	93 Np Neptunium 237	94 Pu Plutonium 244	95 Am Americium 243	96 Cm Curium 247	97 Bk Berkelium 247	98 Cf Californium 251	99 Es Einsteinium 252	100 Fm Fermium 257	101 Md Mendelevium 258	102 No Nobelium 259	103 Lr Lawrencium 260	

Note: Elements 113, 115, and 117 are not currently known.

Note: The subgroup numbers 1-10 were adopted in 1988 by the International Union of Pure and Applied Chemistry (IUPAC) to the periodic table in the form of a temporary solution to the ambiguity of elements and groups (see below).

57 La Lanthanum 138.90547	58 Ce Cerium 140.12	59 Pr Praseodymium 140.90766	60 Nd Neodymium 144.24	61 Pm Promethium 144.9127	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92534	66 Dy Dysprosium 162.5	67 Ho Holmium 164.93033	68 Er Erbium 167.257	69 Tm Thulium 168.93048	70 Yb Ytterbium 173.05448	71 Lu Lutetium 174.967
89 Ac Actinium 227	90 Th Thorium 232.0377	91 Pa Protactinium 231.036889	92 U Uranium 238.02891	93 Np Neptunium 237	94 Pu Plutonium 244	95 Am Americium 243	96 Cm Curium 247	97 Bk Berkelium 247	98 Cf Californium 251	99 Es Einsteinium 252	100 Fm Fermium 257	101 Md Mendelevium 258	102 No Nobelium 259	103 Lr Lawrencium 260

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Eric R. Scerri



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