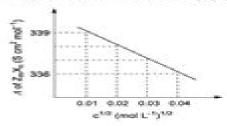


The plot of molar conductivity (A) of Z_nX_n vs c^{1/2} is given below.



Answer (7)

Sol.
$$\lambda_m = \lambda_m^o - A\sqrt{C}$$

For electrolyte Z...X. and from given curve

$$\lambda_m(Z_mX_n) = \lambda_m^o(Z_mX_n) - A\sqrt{C}$$

$$-A = \frac{336 - 339}{0.04 - 0.01} = -\frac{3}{0.03}$$

$$\Rightarrow$$
 A = 100

$$\Rightarrow$$
 336 = $\lambda_m^0 (Z_m X_n) - 100 \times 0.04$

$$\lambda_m^o = 336 + 4 = 340 \text{ S cm}^2 \text{ mol}^{-1}$$

$$Z_m X_n \longrightarrow mZ^{n+} + nX^{m-}$$

$$\implies$$
 5m + 8n = 34

$$U_m Y_p \longrightarrow mU^{p+} + pY^{m-}$$

$$\therefore$$
 25m + 100p = $\lambda_m^{\circ}(U_m Y_p) = 250$

$$\Rightarrow$$
 m + 4p = 10

$$V_m X_n \longrightarrow m V^{n+} + n X^{m-}$$

From equation (i) and (iii)

$$n = 3$$

$$m = 2$$

From equation (ii)

$$p = 2$$

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Joseph R. Stetter, William Roy Penrose

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Solution Chemistry of Surfactants K.L. Mittal, 2012-12-06 The 52nd Colloid and Surface Science Symposium of the Divis ion of Colloid and Surface Chemistry of the American Chemical Society was held in Knoxville TN June 12 14 1978 and one of its Sections was devoted to the topic of Solution Chemistry of Surfactants Although it was billed as the Section on Solution Chemistry of Surfactants but it was indeed a veritable international symposium on this topic as 51 papers by about 100 con tributors from 12 countries were listed in the program The present volume and its companion volume 2 document the proceedings of the above mentioned Section on Solution Chemistry of Surfactants In 1976 there was held an international symposium on Micellization Solubilization and Microemulsions in Albany I the proceedings of which have been chronicled in two volumes A great deal of material dealing with micelles contributed by a legion of prominent researchers constitutes these volumes but a few subtopics were not adequately covered so it was deemed appropriate to cover these topics as well as the recent progress in the general area of aggregation of surfactants in this Section Also as it is the amphiphilicity or amphipathicity of a surfact ant molecule which is responsible for both adsorption at interfaces and aggregation in solution so it was considered guite apropos to include the topic of adsorption at interfaces in this Section Concomitantly the present volumes not only cover the aggregation phenomena but also the adsorption at interfaces Chemistry Leading to Novel Materials Rajendra P. Agarwala, 2001 Recent advances in solid state chemistry have resulted in substantial progress towards achieving a better understanding of the solid state and have even led to the development of new predictive capabilities in crystal chemistry Entirely new ways of studying and preparing advanced materials have been the result of pursuing the so called soft chemistry approach to materials science Soft Chemistry Routes to New Materials Jean Rouxel, Michel Tournoux, R. Brec, 1994 Within the last few decades advances in solid state chemistry have resulted in substantial progress towards a better understanding of the solid state and have even led to the development of some predictive capabilities in crystal chemistry Nonetheless in more recent years entirely new ways of thinking have appeared The term soft chemistry or chimie douce is often applied in a general manner to refer to these new routes these new ways of **Green Solvents for Chemistry** William M. Nelson, 2003-03-27 The aim of this book is to introduce the use of thinking green solvents throughout chemistry and to provide a comprehensive reference for solvents currently applicable in green chemistry The first section covers solvents in chemical perspective and the second section is a guide to green solvents Overall this volume defines characteristics of green solvents and their current usage and explores their importance ecologically and economically It includes a full range of commercial industrial and academic green solvents and discusses solvents in specific commercial and non commercial practices Green Solvents for Chemistry differs from other works on solvents in that only solvents for green chemistry are included along with their chemical properties and toxicological issues

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- 1. Understanding the eBook Roads Advanced Academics Answers Chemistry
 - The Rise of Digital Reading Roads Advanced Academics Answers Chemistry
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Roads Advanced Academics Answers Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Roads Advanced Academics Answers Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Roads Advanced Academics Answers Chemistry
 - Personalized Recommendations
 - Roads Advanced Academics Answers Chemistry User Reviews and Ratings
 - Roads Advanced Academics Answers Chemistry and Bestseller Lists
- 5. Accessing Roads Advanced Academics Answers Chemistry Free and Paid eBooks
 - Roads Advanced Academics Answers Chemistry Public Domain eBooks
 - Roads Advanced Academics Answers Chemistry eBook Subscription Services
 - Roads Advanced Academics Answers Chemistry Budget-Friendly Options

- 6. Navigating Roads Advanced Academics Answers Chemistry eBook Formats
 - o ePub, PDF, MOBI, and More
 - Roads Advanced Academics Answers Chemistry Compatibility with Devices
 - Roads Advanced Academics Answers Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Roads Advanced Academics Answers Chemistry
 - Highlighting and Note-Taking Roads Advanced Academics Answers Chemistry
 - Interactive Elements Roads Advanced Academics Answers Chemistry
- 8. Staying Engaged with Roads Advanced Academics Answers Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Roads Advanced Academics Answers Chemistry
- 9. Balancing eBooks and Physical Books Roads Advanced Academics Answers Chemistry
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Roads Advanced Academics Answers Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Roads Advanced Academics Answers Chemistry
 - Setting Reading Goals Roads Advanced Academics Answers Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Roads Advanced Academics Answers Chemistry
 - Fact-Checking eBook Content of Roads Advanced Academics Answers Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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