

Principles of Chemistry I: ACS Final Review Multiple choice exam Questions

The number of neutrons in the $^{37}_{17}\text{Cl}^-$ ion is:

- A. 17
- B. 20
- C. 21
- D. 37
- B. 20

An Argon atom is isoelectronic with

- A. Cl
- B. Ca
- C. Ti^{4+}
- D. N^{3-}
- C. Ti^{4+}

Which pair of particles has the same number of electrons?

- A. F^- , Mg^{2+}
- B. Ne, Ar
- C. Br^- , Se
- D. Al^{3+} , P^{3-}

A. F^- , Mg^{2+}

Which ion has twenty-six electrons?

- A. Cr^{2+}
- B. Fe^{2+}
- C. Ni^{2+}
- D. Cu^{2+}
- C. Ni^{2+}

Which statement is true?

- A. The nucleus of an atom contains neutrons and electrons.
- B. The atomic number of an element is the number of protons in one atom.

Multiple Choice Physical Chemistry Exam Questions

Disha Experts



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