

1.50

1.50 The viscosity of a certain fluid is 5×10^{-4} poise. Determine its viscosity in both SI and BG units.

From Appendix E, $10^{-1} \frac{\text{N} \cdot \text{s}}{\text{m}^2} = 1 \text{ poise}$. Thus,

$$\mu = (5 \times 10^{-4} \text{ poise}) \cdot \left(10^{-1} \frac{\text{N} \cdot \text{s}}{\text{m}^2} \right) = \underline{5 \times 10^{-5} \frac{\text{N} \cdot \text{s}}{\text{m}^2}}$$

and From Table 1.4

$$\mu = \left(5 \times 10^{-5} \frac{\text{N} \cdot \text{s}}{\text{m}^2} \right) \left(2.089 \times 10^{-2} \frac{\frac{16.5}{\text{ft}^2}}{\frac{\text{N} \cdot \text{s}}{\text{m}^2}} \right) = \underline{10.4 \times 10^{-7} \frac{16.5}{\text{ft}^2}}$$

1.51

1.51 The kinematic viscosity of oxygen at 20°C and a pressure of 150 kPa (abs) is 0.104 stokes. Determine the dynamic viscosity of oxygen at this temperature and pressure.

$$\mu = \nu \rho$$

$$\rho = \frac{p}{RT} = \frac{150 \times 10^3 \frac{\text{N}}{\text{m}^2}}{\left(259.8 \frac{\text{J}}{\text{kg} \cdot \text{K}} \right) \left[(20^\circ\text{C} + 273) \text{K} \right]} = 1.97 \frac{\text{kg}}{\text{m}^3}$$

$$\nu = 0.104 \text{ stokes} = 0.104 \frac{\text{cm}^2}{\text{s}}$$

$$\mu = \left(0.104 \frac{\text{cm}^2}{\text{s}} \right) \left(10^{-4} \frac{\text{m}^2}{\text{cm}^2} \right) \left(1.97 \frac{\text{kg}}{\text{m}^3} \right)$$

$$= 2.05 \times 10^{-5} \frac{\text{kg}}{\text{m} \cdot \text{s}} = \underline{2.05 \times 10^{-5} \frac{\text{N} \cdot \text{s}}{\text{m}^2}}$$

Munson Fundamentals Of Fluid Mechanics Solution Manual 6th Edition

R. Wane Schneiter



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