

Chapter 5 Gases

Student: _____

1. A pressure that will support a column of Hg to a height of 256 mm would support a column of water to what height? The density of mercury is 13.6 g/cm^3 ; the density of water is 1.00 g/cm^3 .
 - A. 348 cm
 - B. $1.00 \times 10^2 \text{ ft}$
 - C. 18.8 mm
 - D. 33.8 ft
 - E. 76.0 cm
2. Which of the following is/are characteristic(s) of gases?
 - A. High compressibility
 - B. Relatively large distances between molecules
 - C. Formation of homogeneous mixtures regardless of the nature of gases
 - D. High compressibility AND relatively large distances between molecules
 - E. High compressibility, relatively large distances between molecules AND formation of homogeneous mixtures regardless of the nature of gases
3. A sample of a gas occupies $1.40 \times 10^3 \text{ mL}$ at 25°C and 760 mmHg. What volume will it occupy at the same temperature and 380 mmHg?
 - A. 2,800 mL
 - B. 2,100 mL
 - C. 1,400 mL
 - D. 1,050 mL
 - E. 700 mL
4. A sample of nitrogen gas has a volume of 32.4 L at 20°C . The gas is heated to 220°C at constant pressure. What is the final volume of nitrogen?
 - A. 60.4 L
 - B. 19.3 L
 - C. 31.4 L
 - D. 54.5 L
 - E. 3.6 L

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