

Problemas resueltos de Química analítica, PQA 4.20.

4.20. Una muestra de sidra de 100.00 g se somete a un proceso de oxidación mediante el cual su contenido en sacarina es transformado en sulfato, que posteriormente es precipitado con bario. Si la masa de este precipitado es de 0.0223 g, calcular el porcentaje de sacarina en la muestra de sidra

La sacarina es oxidada a sulfato y el sulfato es precipitado con bario según las reacciones siguientes y relaciones estequiométricas:



$$\text{moles sacarina/moles S} = \text{moles SO}_4^{2-} = \text{moles BaSO}_4$$

$$\text{moles sacarina} = \text{g BaSO}_4 / \text{Pm BaSO}_4 = 0.0223 / 233.40 = 9.55 \cdot 10^{-5}$$

$$\text{g sacarina} = 9.55 \cdot 10^{-5} \cdot \text{Pm sacarina} = 9.55 \cdot 10^{-5} \cdot 183.19 = 0.0175 \text{ g}$$

$$\% \text{ sacarina} = 0.0175 \cdot 100 / 100.00 = 0.0175\%$$

Problemas resueltos de Química analítica, PQA 4.19.

4.19. Una muestra de 1.227 g de alumbre, $\text{K}_2\text{SO}_4 \cdot \text{Al}_2(\text{SO}_4)_3 \cdot 24 \text{H}_2\text{O}$, que contiene sólo impurezas inertes, se disuelve y el aluminio se precipita con 8-hidroxiquinoleína. Una vez filtrado, el precipitado se calcina a Al_2O_3 , obteniéndose una masa de 0.1098 g. Calcular los porcentajes de aluminio y azufre en la muestra.

El aluminio precipita con la 8-hidroxiquinoleína $\text{C}_9\text{H}_7\text{NO}$, según la reacción:



Una vez calcinado a Al_2O_3 y determinado su peso, 0.1098 g, se puede calcular el contenido de aluminio en la muestra.

$$\text{g Al} = \text{g Al}_2\text{O}_3 \cdot 2 \cdot \text{Pat Al} / \text{Pm Al}_2\text{O}_3 =$$
$$= 0.1098 \cdot 2 \cdot 26.98 / 101.96 = 0.05811 \text{ g}$$

$$\% \text{ Al} = 0.05811 \cdot 100 / 1.2277 = 4.73\%$$

Para calcular el contenido de azufre en peso y su porcentaje tendremos en cuenta la reacción estequiométrica:

Problemas Resueltos De Quimica Analitica

LOPEZ CANCIO, JOSE ANTONIO

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Problemas Resueltos De Química Analítica:

Problemas resueltos de Química Analítica Cuantitativa Jorge Suárez Sandoval, 2019-10-08 El Ingeniero Víctor H Montenegro Velázquez catedrático de Química Analítica en la Universidad Nacional de la Amazona Peruana UNAP en su Manual de Prácticas de Química Analítica Cuantitativa plantea 201 Problemas de Química Analítica Cuantitativa para entrenar a sus alumnos no solamente en la técnica analítica sino también en el manejo e interpretación de los resultados obtenidos Este es el origen del presente libro publicado en mayo de 1968 en la ciudad de Iquitos que hoy cincuenta años después he decidido volver a publicarlo en formato digital porque aunque el análisis químico instrumental ha avanzado mucho todavía los métodos volumétricos siguen teniendo vigencia y se hace por tanto necesario su manejo Es un texto para aquellos estudiantes que tienen que ver con Laboratorios de Análisis Químico sea cual sea su campo Ser de mucha utilidad en la determinación del método y el análisis de sus resultados y ha de ser también un gran apoyo para los profesionales como manual de consulta permanente para mantenerse en la práctica de estos métodos Problemas resueltos de química analítica (Acceso) JOSE ANTONIO LOPEZ CANCIO, 2005-06-23 **Problemas resueltos de química analítica** LOPEZ CANCIO, JOSE ANTONIO, 2005-01-01 Esta obra pretende cubrir un curso básico de química analítica desde una perspectiva eminentemente práctica El objetivo es que el alumno adquiera una formación básica sobre los equilibrios iónicos básicos en disolución así como sus principales aplicaciones gravimétricas y volumétricas **300 Problemas resueltos de química analítica** Paloma Yáñez-Sedeño Orive, José M. Pingarrón Carrazón, Francisco Javier Manuel de Villena Rueda, 2022 **Química Analítica Básica**, **Problemas resueltos de química analítica** Paloma Yáñez-Sedeño Orive, José Manuel Pingarrón Carrazón, Francisco Javier Manuel de Villena Rueda, 2003 Fundamentos de química analítica. Volumen 2 Douglas A. Skoog, Donald M. West, James Holler, 2020-01-10 Constituye la versión castellana de la última edición 6a de la conocida obra de los profesores Skoog y West enriquecida y actualizada recientemente por el Profesor Holler Por la selección de temas claridad de exposición y enfoque moderno sigue siendo un texto recomendable como introducción a la Química analítica para estudiantes no solo de Química o Farmacia sino también y en general de carreras técnicas que incluyan la Química en sus planes de estudios **Química analítica cualitativa (reacciones en solución)** Vicente Bolaños Chombo, 2003 *Análisis Instrumental* Isabel Sierra Alonso, Santiago Gómez Ruiz, Damián Pérez Quintanilla, Sonia Morante Zarcero, 2009-08 Entre las implicaciones que supone la adaptación de las actuales titulaciones al Espacio Europeo de Educación Superior EEES destaca el diseño de guías docentes para las asignaturas basadas en competencias y objetivos de aprendizaje Este hecho afecta tanto a las metodologías de enseñanza aprendizaje necesarias para la adquisición de dichas competencias como al sistema de evaluación de las mismas Por ello con objeto de adaptar la enseñanza en las universidades españolas a este nuevo espacio es necesario reflexionar sobre algunas cuestiones relacionadas con los cambios metodológicos y de evaluación El libro que se presenta supone una innovación respecto a los tradicionalmente utilizados para el estudio del Análisis Instrumental ya que con

l se apuesta por un modelo de aprendizaje centrado en el alumno y una formaci n basada en competencias De esta forma el libro ha sido concebido para ofrecer al estudiante y al profesor materiales y recursos que permitan la ense anza aprendizaje de esta materia en el nuevo marco del EEES Las actividades presentadas est n encaminadas al desarrollo evaluaci n de competencias gen ricas especialmente valoradas en estudiantes de Grados Cient fico Tecnol gicos favoreci ndose el aprendizaje aut nomo por parte del mismo as como su participaci n activa en el desarrollo de la asignatura Problemas Resueltos Manuel Castro Vicente, Jes s Silva Valdivia, 2003

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