(a) You are provided with an organic compound C-13. Perform the experiments given below on the compound. Record the changes taking place at each step of the experiment.

Note the smell of the substance formed, if significant, the colour of the solution obtained, the colour of the precipitate produced and any other observations you may have. State the identity of the compound on the basis of the experiments and observational changes.

Substance C-13

PROCEDURE:

- Take 1 ml of C-13 in a test tube and add a few drops of water. Add 2 3 drops of sodium hypochlorite solution and shake the contents.
- (ii) Take 1 ml of C-13 in a test tube and add 1 ml of concentrated hydrochloric acid to it. Now, add a few drops of neutral ferrie chloride solution and dilute the contents with water.
- (iii) Take I ml of C-13 in a test tube and add a few drops of dilute sulphuric acid. Now, add I ml of potassium dichromate solution. Shake and warm the contents.
- (b) Substance C-14 is an unknown sample of either carbohydrate or protein. Carry out the following experiments and record all your observations. State the identity of the compound as carbohydrate or protein on the basis of the experiments and observational changes.

Substance C-14

PROCEDURE:

Take the sample C-14 in a test tube. Dissolve it in 10 ml of distilled water in order to obtain saturated solution. Divide the solution into three parts.

- To the first part of C-14, add 1 ml of Fehling's solution. Warm the contents.
- (ii) To the second part of C-14, add 2 3 drops of ar naphthol solution. Now, pour concentrated sulphuric acid slowly along the sides of the test tube.
- (iii) To the third part of C-14, add 1 ml of lead acetate solution followed by 1 ml ammonium hydroxide solution. Boil the contents.

3

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