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Hermann Sicius

Seltenerdmetalle: Lanthanoide und dritte Nebengruppe

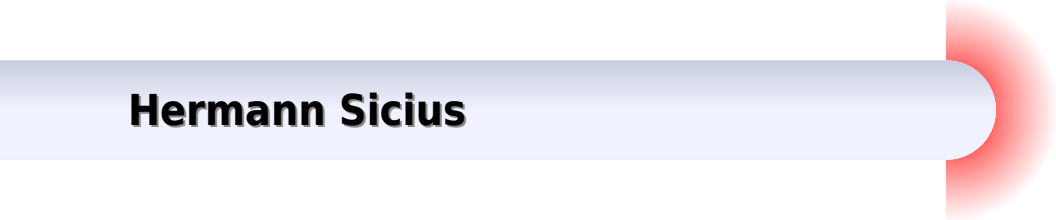
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Seltenerdmetalle Lanthanoide Dritte Nebengruppe Essentials

Hermann Sicius



Seltenerdmetalle Lanthanoide Dritte Nebengruppe Essentials:

Seltenerdmetalle: Lanthanoide und dritte Nebengruppe Hermann Sicius, 2015-07-01 Hermann Sicius bringt dem Leser in knapper Form alle wichtigen Informationen rund um eine wichtige Klasse metallischer Rohstoffe nahe. Die insgesamt achtzehn Metalle der Lanthanoide Seltenerdmetalle sowie der dritten Nebengruppe sind nur einem kleinen Teil der Öffentlichkeit bekannt, aber von großer Bedeutung für viele zukünftige Technologien.

Kupfergruppe: Elemente der ersten Nebengruppe Hermann Sicius, 2017-01-27 Hermann Sicius stellt ausführlich die Elemente der ersten Nebengruppe Kupfer Silber Gold Roentgenium vor, deren physikalische und chemische Eigenschaften relativ ähnlich sind. Kupfer Silber und Gold sind bereits seit Jahrtausenden bekannt. Beim Elementenpaar Silber und Gold ist die Lanthanoidenkontraktion nur noch abgeschwächt erkennbar. Die jeweiligen physikalischen Eigenschaften von Kupfer Silber und Gold hängen sich bis auf die von Kupfer zu Gold zunehmende Dichte dafür nimmt wie auch schon bei den Elementen der vierten bis zehnten Nebengruppe die Reaktivität vom Kopfelement hier Kupfer zum schwersten Gruppenmitglied hier Gold zu. Auch Kupfer ist aber schon ein Halbedelmetall. Bei Kupfer sind die Oxidationsstufen 1 und 2 am stabilsten, bei Silber 1 und bei Gold 1 und 3. Erst 1994 konnte das erste Atom des Roentgeniums erzeugt werden.

Zinkgruppe: Elemente der zweiten Nebengruppe Hermann Sicius, 2017-04-24 Hermann Sicius stellt ausführlich die Elemente der zweiten Nebengruppe vor: Zink Cadmium Quecksilber Copernicium, deren physikalische und chemische Eigenschaften relativ ähnlich sind. Auswirkungen der Lanthanoidenkontraktion erkennt man hier kaum. Cadmium steht in seinen Eigenschaften etwa zwischen dem Zink und dem Quecksilber. Zink und Cadmium haben negative Normalpotentiale, wogegen Quecksilber ein Halbedelmetall ist. Die Elemente dieser Gruppe geben meist ein oder zwei äußere Valenzelektronen ab, um eine stabile Elektronenkonfiguration zu erreichen. Für das höchste Element dieser Nebengruppe, das Copernicium, konnten bisher kaum chemische Untersuchungen durchgeführt werden. Es ist zu erwarten, dass es sich chemisch ähnlich wie Quecksilber verhält. Zink als Element kennt man seit dem 17. Jahrhundert, Cadmium seit 1817, wogegen Quecksilber schon in der Antike bekannt war. Die erstmalige Darstellung von Atomen des Coperniciums gelang 1996.

Vanadiumgruppe: Elemente der fünften Nebengruppe Hermann Sicius, 2016-06-14 Hermann Sicius stellt in diesem Essential die Elemente der fünften Nebengruppe, Vanadiumgruppe, ausführlich vor. Er zeigt, dass diese Elemente teilweise reaktiv, aber gelegentlich auch sehr reaktionsträge sind, dass sie an der Luft lagernd von einer dicken passivierenden Oxidschicht vor weiterer Korrosion geschützt und auch in Säuren nur selten und dann nur unter Anwendung drastischer Methoden löslich sind. Die technische Anwendung zeigt der Autor für diese Elemente auf: die alle sehr harte, hochschmelzenden Metalle sind und deren Verbindungen mit Kohlen und Stickstoff teils extrem hochschmelzende, harte Schleifmaterialien darstellen, die man auch industriell einsetzt. Vanadium findet sich in Stahlliegierungen, Tantal in medizinischen Implantaten, Ferroniob als Bestandteil von Gasturbinen oder Vanadium Voxid als Katalysator bei der Herstellung von Schwefelsäure.

Chromgruppe: Elemente der sechsten Nebengruppe Hermann

Sicius, 2016-06-23 Hermann Sicius stellt in diesem essential die Elemente der sechsten Nebengruppe Chromgruppe ausführlich vor. Er zeigt, dass die Elemente dieser Gruppe chemisch meist relativ reaktionsträge sind und regelmäßige Abstufungen ihrer Eigenschaften zeigen. So nehmen vom Chrom zum Wolfram Dichte, Schmelz- und Siedepunkte zu, die Reaktivität leicht ab. Chrom, Molybdän und Wolfram entdeckte man Ende des 18. Jahrhunderts; das nur künstlich darstellbare Seaborgium erst etwa 200 Jahre später. Molybdän ist z.B. Bestandteil sehr harter Stähle. Wolfram wurde für Wendeln von Glühlampen verwendet.

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