

Unit 3 Study Guide

Be sure to review the Unit 3 Module on Canvas, including the notes, and any supplemental pages.

The following are items that will appear on your Unit 3 test. These are practice questions that may or may not appear on the test. This should be used as a guide to study for your test and doesn't include EVERYTHING we covered during this unit.

(However, students who are able to answer these questions on their own or with little help will be well-prepared for the test.)

****There was a lot of information during this unit, so please study! If you have questions, please sign up for my Encore! I will be holding a review session before the test!**

Answer the following questions on a separate piece of paper.

1. The first idea of the atom was suggested by a philosopher, Democritus, in 460 BCE. Why was he not credited with the discovery of the atom?

2. Models of the Atom (Dalton, Thomson, Rutherford, Bohr, Schrodinger)

- What did each model show? (Sketch it out!)
- Which parts of the model were accurate and which parts were not?

3. Fill in the missing information of subatomic particles:

Proton			
		Negative (-)	
	1 atom		in the nucleus

4. What is the purpose of the neutron in the nucleus?

5. What is an isotope? (Which subatomic particle number changes?) Write out an example.

6. How many protons, neutrons, and electrons does this isotope have? Chlorine-37

7. What is an ion? (Which subatomic particle number changes?) Write out an example.

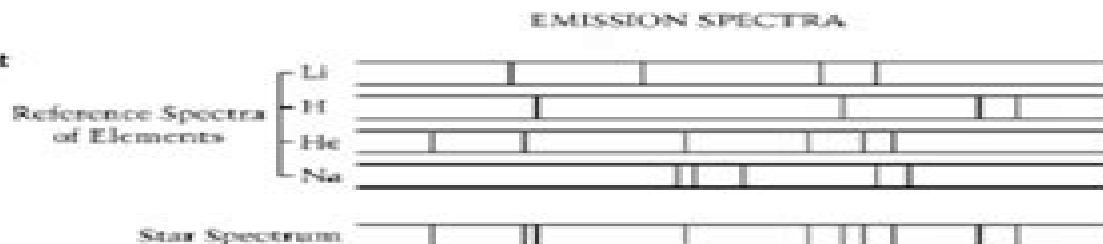
8. How many protons, neutrons, and electrons does this ion have? Al^{3+}

9. Explain what the atomic mass reported on the periodic table actually tells us.

10. Typically, elements larger than iron can not fuse together. Explain what happens in a supernova that allows larger elements to form.

11. Explain nucleosynthesis and sketch out an example to show how nuclei fuse to form larger elements.

12. Use the emission spectra to the right, to determine the 2 elements that make up the unknown star.



Separate Piece Study Guide Answer

E Durkheim



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excerpted from Gale's acclaimed *Novels for Students*. This concise study guide includes plot summary, character analysis, author biography, study questions, historical context, suggestions for further reading, and much more. For any literature project, trust *Novels for Students* for all of your research needs. [Understanding A Separate Peace](#) Hallman Bryant, 2001-11-30

Since its publication in 1959, *A Separate Peace* has acquired the reputation of a minor classic of American literature. This insightful analysis helps young readers relate to the themes of disillusionment, guilt, and betrayal, and the fear of failure and intergenerational conflicts experienced by the teenaged characters in the novel. This casebook also situates *A Separate Peace* against the backdrop of World War II, enabling students to see the connections between the fictional world of the novel and the real world as it existed for young people. Moving well beyond a standard literary treatment, this interdisciplinary casebook provides a collection of historical primary documents drawn from official records, War Department orders, institutional histories, personal memoirs, and letters, and poignant interviews. With commentary by Knowles himself, the casebook takes readers from the prep school setting of the novel to the impact of wartime on American students and their schools. *You're in the Army Now* explores the difficult transitions through induction and military training. *The Combat Zone* graphically confronts the realities of war with interviews of two former POWs who experienced firsthand the terrors and tragedies of WWII. The volume also examines some of the contemporary issues of the novel, including current controversies in athletic programs, gender issues in education, and Post-Traumatic Stress Disorder. Teachers and librarians will find helpful suggestions for oral discussion, research projects, and further suggested readings on these important topics. [Deeper Reading](#) Kelly Gallagher, 2023-10-10

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