Chemistry Honors Final Study Guide

Solids, Liquids, and Solutions | Chapters 14 & 15:

Intermolecular Forces

- Intermolecular Forces: forces that hold molecules together in liquids and solids (are weaker than intramolecular forces)
 - Types of Intramolecular forces are Ionic Bonds, covalent bonds, and metallic bonds.
 - The 3 intermolecular forces are dipole-dipole forces, dispersion forces, and hydrogen banding.

. Dipole - Dipole Forces

- Dipole-Dipole Forces: attractions between opposite charges of neighboring (permanent) dipoles.
- Polar molecules are attracted to opposite ends of each other through dipole-dipole forces.
- · This intermolecular force is only possible with polar molecules

. Ion-Dipole Forces

Ion Dipole Force: an attractive force that results from the electrostatic attraction between an ion
and a neutral molecule that has a dipole

. Dispersion Forces

- Induced Dipole: a dipole created by the presence of a neighboring dipole
- Dispersion Forces: the force of attraction between induced dipoles (occur whenever electrons are involved)
- Polarizability: the ease with which the electron cloud of an atom or molecule is distorted
 - The respective trend is atomic (or molar) mass/radius/weight and it is the exact opposite of the ΔEN trend on the periodic table

Semester 2 Chemistry Honors Review Guide

R Sandford

Semester 2 Chemistry Honors Review Guide:

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 - Chapter 1: Introduction to Semester 2 Chemistry Honors Review Guide
 - o Chapter 2: Essential Elements of Semester 2 Chemistry Honors Review Guide
 - \circ Chapter 3: Semester 2 Chemistry Honors Review Guide in Everyday Life
 - Chapter 4: Semester 2 Chemistry Honors Review Guide in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, this book will delve into the foundational concepts of Semester 2 Chemistry Honors Review Guide. The second chapter will elucidate the essential principles that need to be understood to grasp Semester 2 Chemistry Honors Review Guide in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Semester 2 Chemistry Honors Review Guide in daily life. This chapter will showcase real-world examples of how Semester 2 Chemistry Honors Review Guide can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of Semester 2 Chemistry Honors Review Guide in specific contexts. This chapter will explore how Semester 2 Chemistry Honors Review Guide is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Semester 2 Chemistry Honors Review Guide. The final chapter will summarize the key points that have been discussed throughout the book.
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