

CHEMISTRY



Periodic Table of Elements

Properties of Matter

Matter is anything that has mass and takes up space. It can be solid, liquid, or gas.

States of Matter

- Solid:** Particles are packed closely together and vibrate in place.
- Liquid:** Particles are close together but can move past each other.
- Gas:** Particles are far apart and move rapidly in all directions.

Changes of State

- Melting:** Solid to liquid.
- Freezing:** Liquid to solid.
- Boiling:** Liquid to gas.
- Condensation:** Gas to liquid.
- Sublimation:** Solid to gas.
- Deposition:** Gas to solid.

Atomic Structure

Atoms are made of three subatomic particles: electrons, protons, and neutrons.

Electrons: Negatively charged particles found in all atoms.

Protons: Positively charged particles found in the nucleus.

Neutrons: Neutral particles found in the nucleus.

Atomic Structure

Atoms are the smallest particles of an element that can take part in a chemical reaction. They are made up of three subatomic particles: electrons, protons, and neutrons.

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Atomic Structure

Element	Atomic Number	Mass Number
Hydrogen	1	1
Helium	2	4
Lithium	3	7
Boron	5	11
Carbon	6	12
Nitrogen	7	14
Oxygen	8	16
Fluorine	9	19
Neon	10	20

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