

### 3.ACID,BASES AND SALT

#### Intext Questions 1

##### Question 1

- (a) What do you understand by the term acid?  
 (b) Name the positive ion formed when an acid is dissolved in water.  
 (c) Draw the structure of this ion.

**Answer**

(a) Acids are defined as compounds which contain one or more hydrogen atoms, and when dissolved in water, they produce hydronium ions ( $\text{H}_3\text{O}^+$ ) as the only positively charged ions.

(b) Hydronium ion ( $\text{H}_3\text{O}^+$ )

(c) The structure of hydronium ion is shown below:



##### Question 2

Write the ionisation of sulphuric acid showing the formation of hydronium ion.

**Answer**



##### Question 3

Water is never added to acid in order to dilute it. Why?

**Answer**

Water is not added to acid as it is an exothermic process and in this process so much heat is produced that splashing of acidic acid solution may occur, also the container may break which can be fatal to the person.

##### Question 4

Define the term 'basicity' of an acid. Give the basicity of: nitric acid, sulphuric acid and phosphoric acid.

**Answer**

The basicity of an acid is defined as the number of hydronium ions ( $\text{H}_3\text{O}^+$ ) that can be produced by the ionization of one molecule of that acid in aqueous solution.

The basicity of following compounds are:

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